



# *Hydrogen Production and Utilization*

*A Canadian Perspective - Sam Suppiah*

*Canadian Nuclear Laboratories, Chalk River, Canada*

*Pathways to Decarbonization: An International Workshop to Explore  
Synergies Between Nuclear and Renewable Energy Sources*

*National Renewable Energy Laboratory, Golden, Colorado*

*2016 Jun 9-10*



Canadian Nuclear Laboratories | Laboratoires Nucléaires Canadiens

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# *Presentation Outline*

- **Introduction**
  - **Hydrogen demand**
  - **Hydrogen & Canada**
  - **Hydrogen for transportation**
- **Hydrogen Production and Utilization at Canadian Nuclear Laboratories (CNL)**
  - **Hydrogen Production from Nuclear Technology**
  - **Hydrogen Utilization**
- **Hydrogen Energy Storage**
- **Energy Scene in Canada – Nuclear and Renewables**



# Hydrogen Demand

- Current world demand for H<sub>2</sub> : ~50-60 Mt/a (increasing rapidly – **USD 118B in 2016 to 152B by 2021**)
  - Ammonia production 40 – 45 Mt/a
  - Methanol 1 – 2 Mt/a
  - Oil refining 10 – 15 Mt/a (growth area)
- H<sub>2</sub> used for synthetic crude upgrading (Canada)  
(2.4 – 4.3 kg H<sub>2</sub> per barrel of bitumen)
  - Current: 2.0 Mt/a
  - By 2020: 6.0 Mt/a (?)
- Hydrogen as a transportation fuel
  - ? Mt/a



# Hydrogen & Canada

- Canada has been a leader in development of hydrogen technologies stretching back for decades. Among best known:
  - Ballard (PEM fuel cells)
  - Hydrogenics Energy Systems
  - Dynatek Industries (high-pressure storage tanks)
- Canada is also a huge hydrogen producer/consumer
  - Nearly three million tonnes of hydrogen per year
    - one-third of the United States' production
    - largest per-capita hydrogen producer in the OECD
    - Oilsands projects are voracious consumers of hydrogen

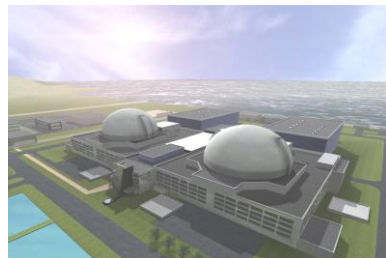


# Hydrogen for Transportation



**Uranium Mining**

+



**CANDU & SCWR**

+



**Electrolysis**

+



**Distribution System**

+



**Fuel Cells**

=

**Power for hydrogen vehicles that could  
replace many gas burning vehicles in  
Canada & elsewhere**

**With the benefit of no carbon dioxide  
emissions!**



*Innovative Environmental Solution*

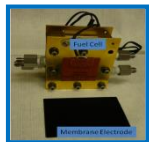




# Hydrogen Production & Utilization at CNL

## Catalyst Production

- Fuel cell
- Electrolysis cell
- Hydrogen Safety Systems
- Hydrogen Isotope Exchange
- Dissolved Oxygen Removal
- Hydrogen Purifier



Fuel Cell Catalyst



Hydrophobic Screen Catalyst



PEM Electrolyser Catalyst

## Membrane and Membrane Electrode Assembly

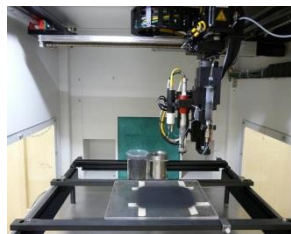
- Fuel Cell
- Electrolysis Cell
- Radiation Resistant



Casting Table & Oven for Membrane Production



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Programmable X-Y-Z-θ Spray Coating Machine



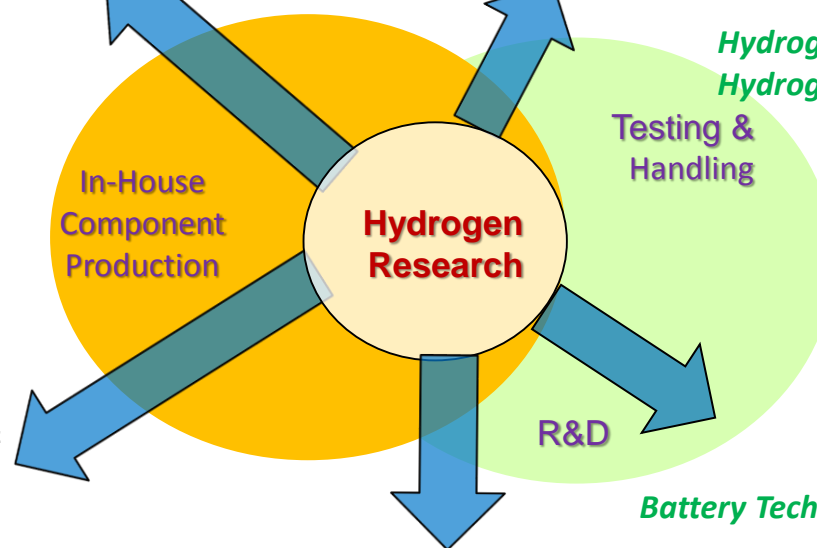
Large-Scale Vented Combustion Test Facility



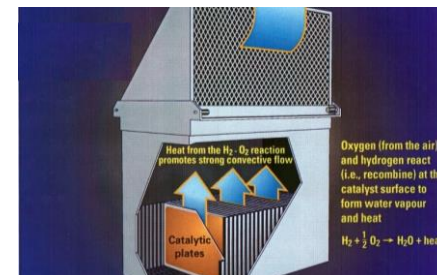
Containment Test Facility



Winnipeg Transit Garage

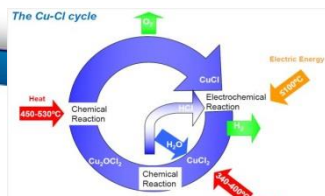


## Hydrogen Safety Systems – Catalytic Hydrogen Recombiners

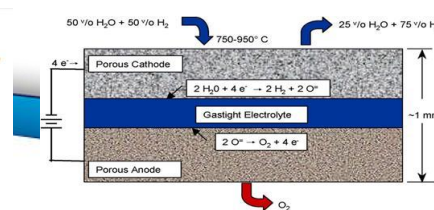


Fuel Cell Test Stations

## Hydrogen Production – Nuclear & Renewable



Copper-Chlorine Thermochemical Cycle



High Temperature Steam Electrolysis

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Metal Evaporator

# Hydrogen from Nuclear Technology

- **Central Issues:**

- **Is it price competitive?**

- ✓ Current status - must use intermittent production at off-peak electricity prices
    - ✓ Fits well with nuclear base-load operation

- **Will the price be stable?**

- ✓ Yes

- **Is it environmentally friendly?**

- ✓ Avoids 8 kg CO<sub>2</sub> per kg of H<sub>2</sub> produced (compared to Steam Methane Reforming)
    - ✓ One 250,000 bbl/d upgrader –save 2.5 Mt CO<sub>2</sub>/a

- **Can production achieve continuity of supply?**

- H<sub>2</sub> storage in underground caverns
    - ICI has used caverns at Teesside UK for 30 years
    - embed in a large supply network



# *Current Nuclear Hydrogen Production Option*

- **Low Temperature Electrolysis - Off-peak electricity storage scenario**
  - Nuclear plants are not designed to load-follow, storage helps in maintaining base-load operation
  - Improves economics of nuclear plant operation
  - Provides large-scale electricity storage capacity
  - Two marketable products: hydrogen and electricity



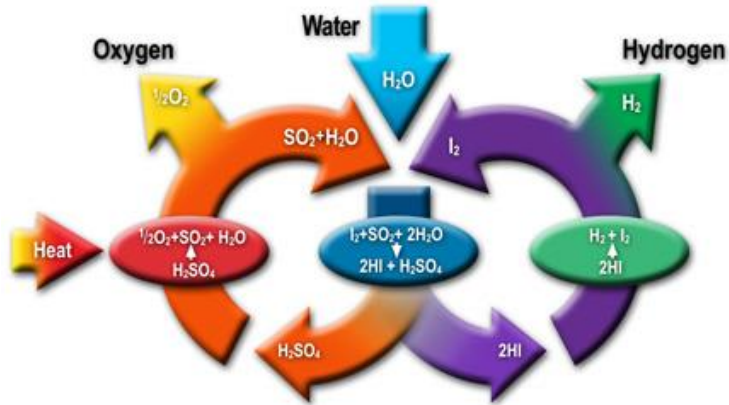


# *Next Generation Nuclear Hydrogen Production Options (also suitable for integration with Solar)*

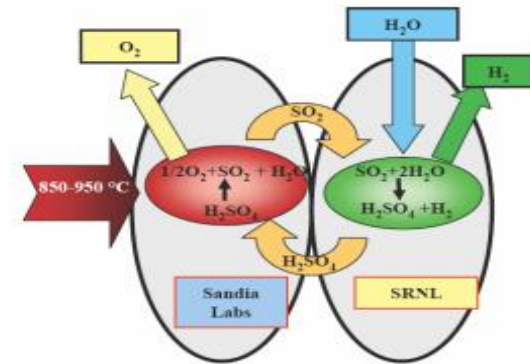
- High Temperature Steam Electrolysis (HTSE)
  - Suitable for integration with Moderate Temperature & High Temperature Nuclear Reactors
  - CO<sub>2</sub> co-electrolysis to produce CO+H<sub>2</sub> (synthetic gas)
- Thermochemical Cycles
  - Sulphur-Iodine (S-I) Process
    - Need Very High Temperature Reactor (Japan, Korea, China, India & others)
  - Hybrid Sulphur (Hyb-S) Process
    - Need Very High Temperature Reactor (US?)
  - Copper Chlorine Process
    - Moderate T Reactors (SCWR, Liquid Metal Reactors, Molten Salt Reactors)
    - Being developed in Canada for integration with Super Critical Water Reactor



# Next Generation Nuclear Hydrogen Production Options

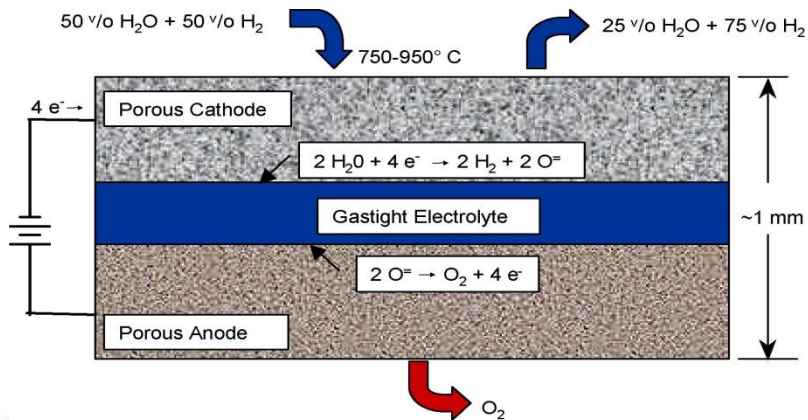


Sulfur-Iodine Process



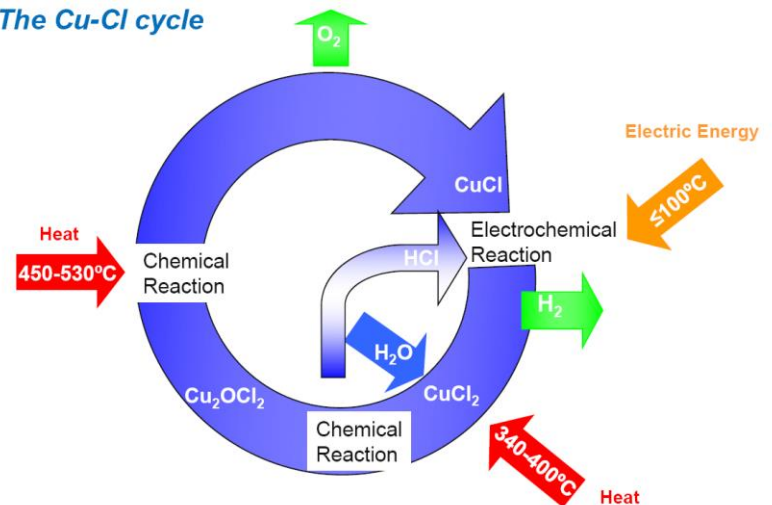
Hybrid-Sulfur

- (1)  $\text{H}_2\text{SO}_4 \rightarrow \text{H}_2\text{O} + \text{SO}_2 + \frac{1}{2}\text{O}_2$
- (2)  $2\text{H}_2\text{O} + \text{SO}_2 \rightarrow \text{H}_2\text{SO}_4 + \text{H}_2$



High Temperature Steam Electrolysis

The Cu-Cl cycle



## *Attractiveness of Cu-Cl Cycle*

- High efficiency (from Scoping Flowsheet Methodology) - 41%
- Low temperature requirement for heat source  $<530^{\circ}\text{C}$
- Temperature requirement for heat source met by currently existing power plant technology (e.g. thermal stations using supercritical water cycles) – BOP technologies exist
- Materials-of-construction and corrosion issues more tractable at  $500^{\circ}\text{C}$  than at higher temperatures required by other cycles
- Inexpensive raw materials as recycle agents (for example, compared to iodine for S-I cycles)
- No requirement for catalyst in thermal reactions



# *Canadian and International Collaborators*

Canadian Nuclear Laboratories (CNL)

University of Ontario Institute of Technology  
(UOIT)

- Faculty of Engineering and Applied Science
- Faculty of Science
- Faculty of Energy Systems & Nuclear Science

University of Toronto

University of Guelph

University of Western Ontario

University of Waterloo

McMaster University

Argonne National Laboratory

Pennsylvania State University

University of Maribor

Czech Academy of Science

Ontario Power Generation

Regional Municipality of Durham

Marnoch Thermal Power Inc.

Phoenix Canada Oil Company Ltd.

Generation IV International Forum

International Nuclear Energy Research  
Initiative; I-NERI

University Network of Excellence in Nuclear  
Engineering; UNENE



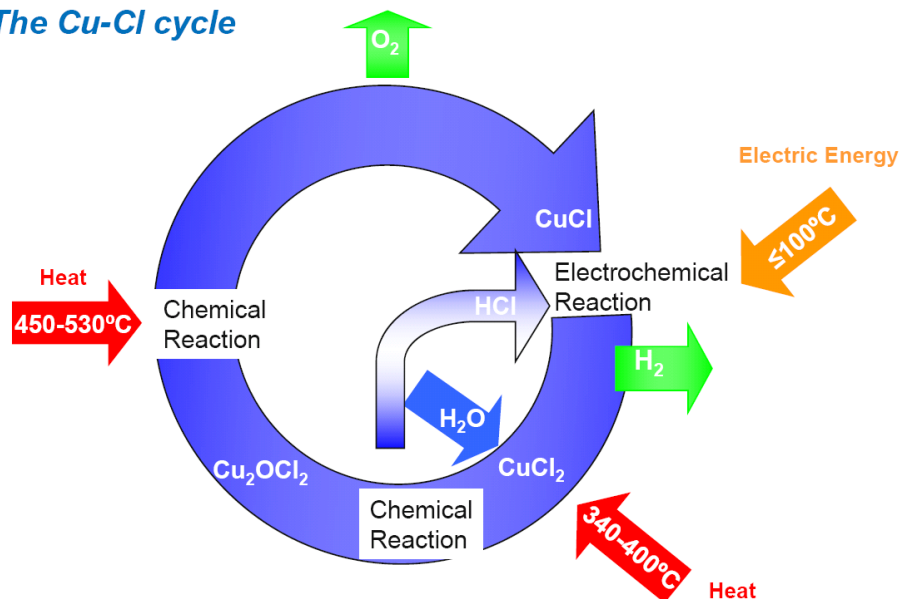


# Copper-Chlorine Thermochemical Cycle (Cu-Cl)

Table The main steps involved in the Copper-Chlorine Cycle.

Step	Reaction	Temperature Range (°C)
1 Electrolysis	$2\text{CuCl}_{(\text{aq})} + 2\text{HCl}_{(\text{aq})} \rightarrow 2\text{CuCl}_{2(\text{aq})} + \text{H}_{2(\text{g})}$	~100
2 Separation/drying	$\text{CuCl}_{2(\text{aq})} \rightarrow \text{CuCl}_{2(\text{s})}$	< 100
3 Hydrolysis	$2\text{CuCl}_{2(\text{s})} + \text{H}_2\text{O}_{(\text{g})} \rightarrow \text{CuO} \cdot \text{CuCl}_{2(\text{s})} + 2\text{HCl}_{(\text{g})}$	350-400
4 Thermal decomposition	$\text{CuO} \cdot \text{CuCl}_{2(\text{s})} \rightarrow 2\text{CuCl}_{(\text{l})} + 1/2\text{O}_{2(\text{g})}$	<b>450-530</b>

## The Cu-Cl cycle

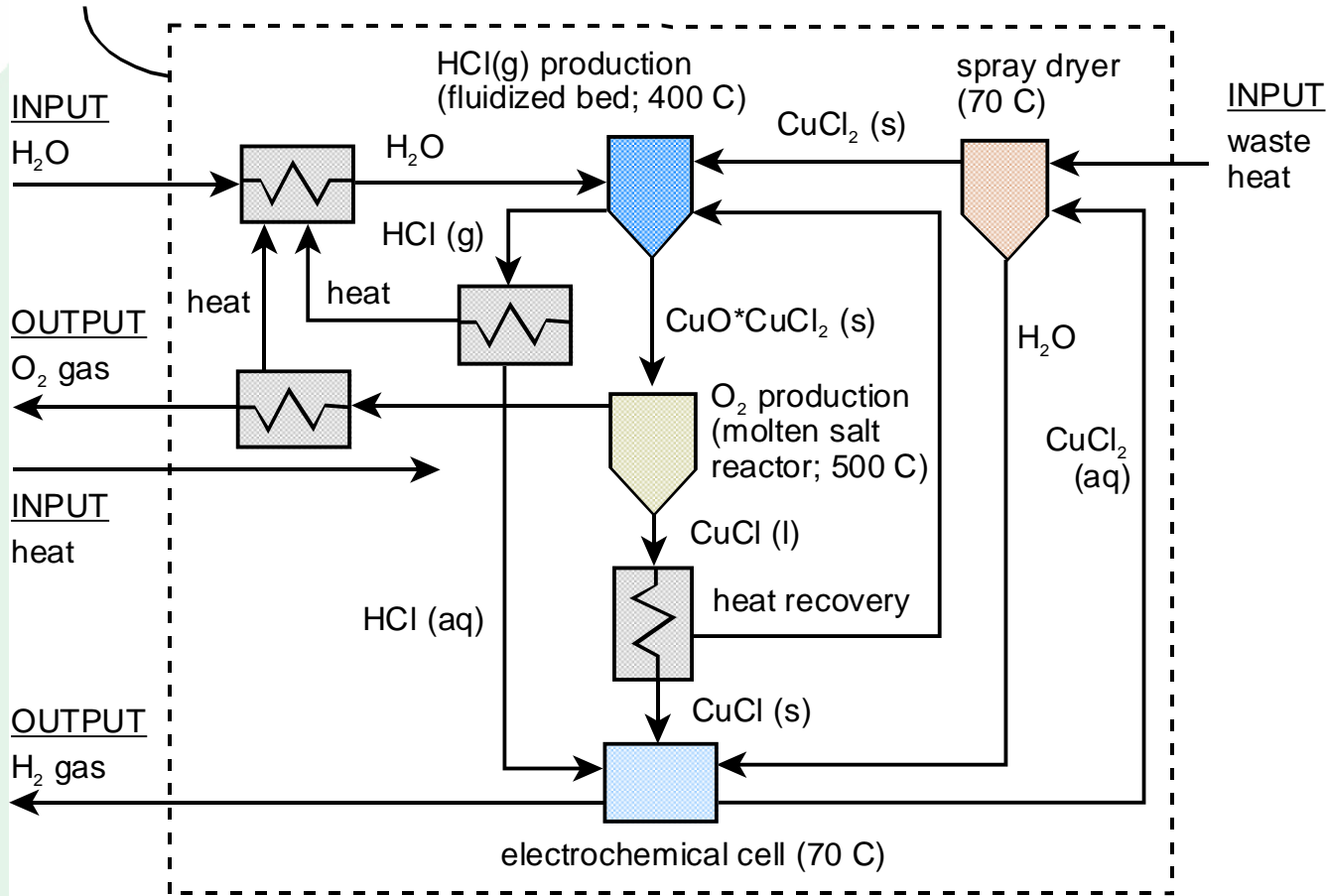


Canadian innovation/demonstration of this step spurred interest – now others following Canadian leadership



# Copper-Chlorine Cycle

Cu-Cl cycle splits  $\text{H}_2\text{O}$  into  $\text{H}_2$  and  $\frac{1}{2}\text{O}_2$  (all other chemicals are re-cycled continuously)



# High Temperature Steam Electrolysis

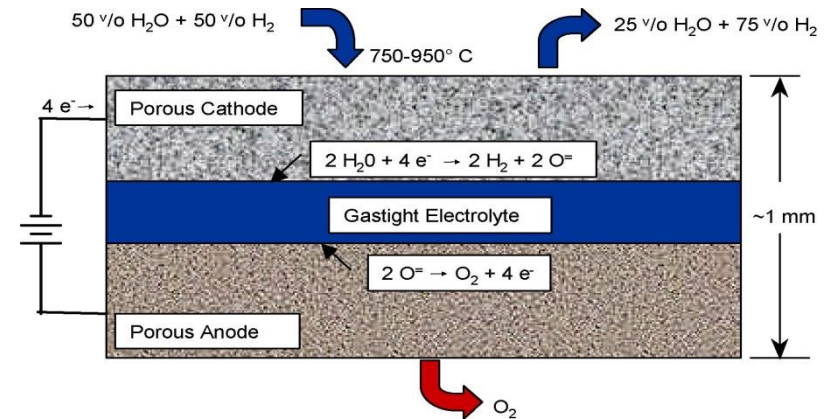
- Canadian efforts:

- Collaborated with INL

- Modelling Integration of HTSE with Nuclear

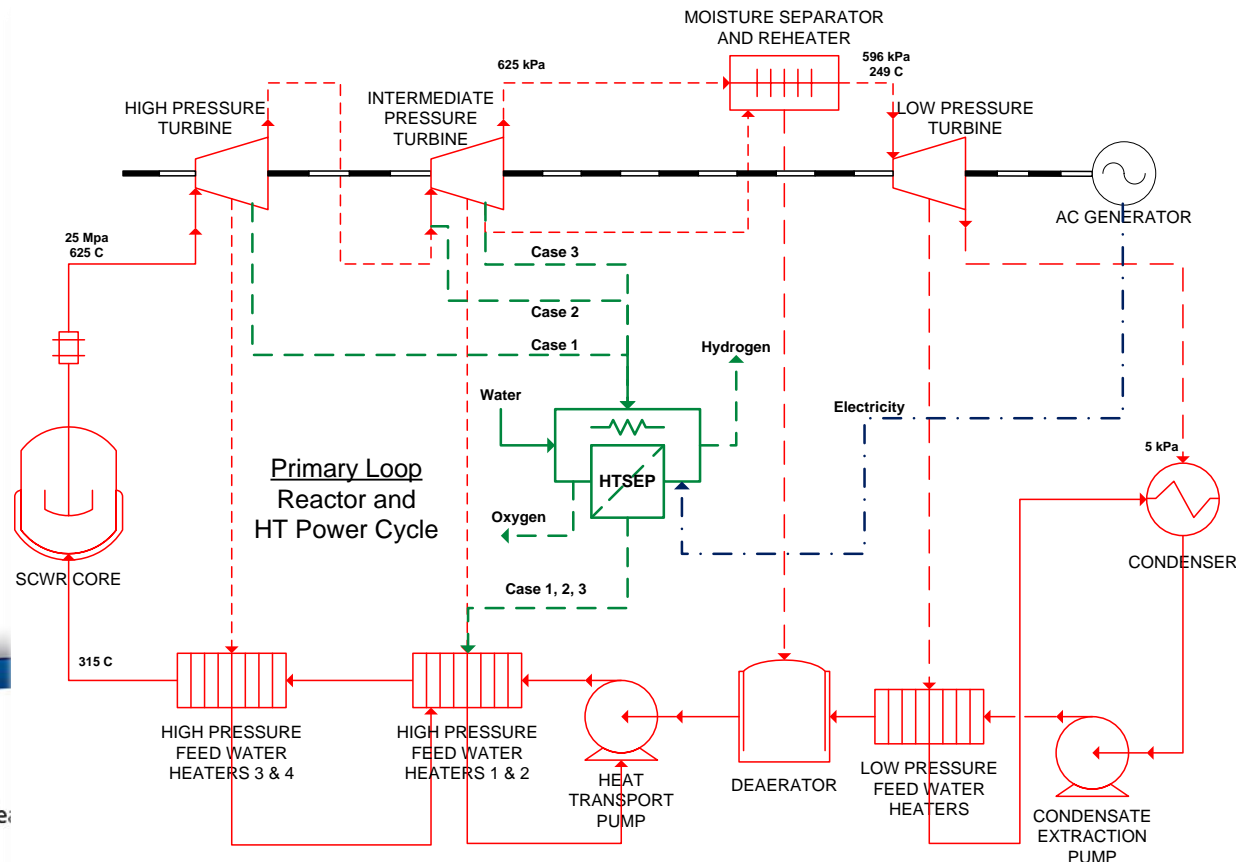
- Materials development

- Focus on lowering the temperature of operation
    - Testing of single cells
    - Testing of a stack of small number of cells



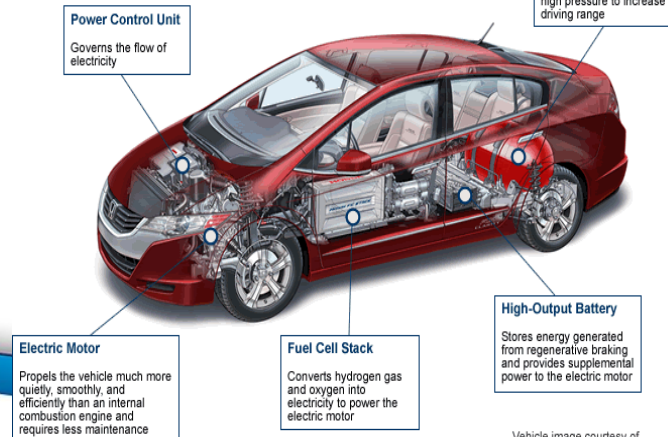
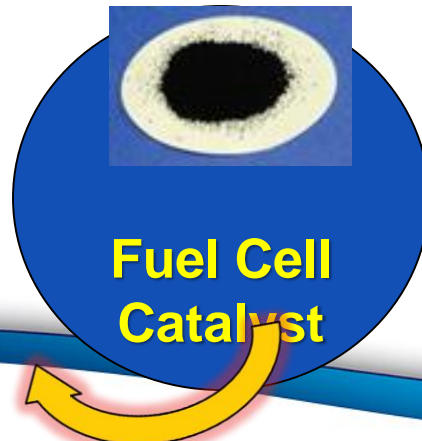
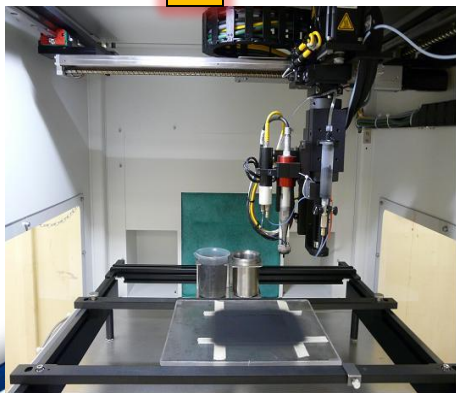
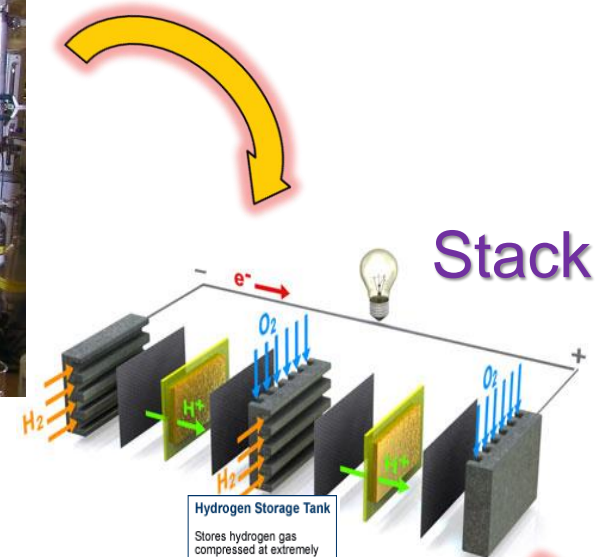
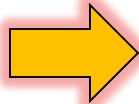
# *Nuclear Hydrogen Production Simulation: HTSE integration with nuclear plants*

- ❑ Simulates hydrogen production from Gen IV reactors
- ❑ Optimizes electrolyser integration with reactor steam cycle
- ❑ Enables assessment for feasibility of large-scale hydrogen production





# Fuel Cell Applications - CNL's catalyst technology



Vehicle image courtesy of American Honda Motor Co., Inc.



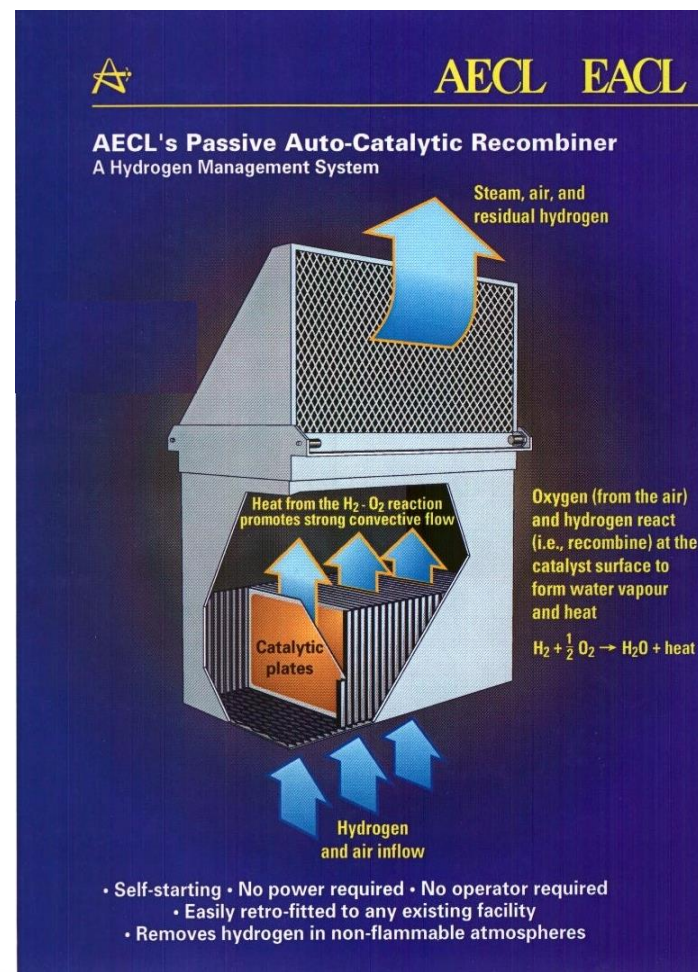
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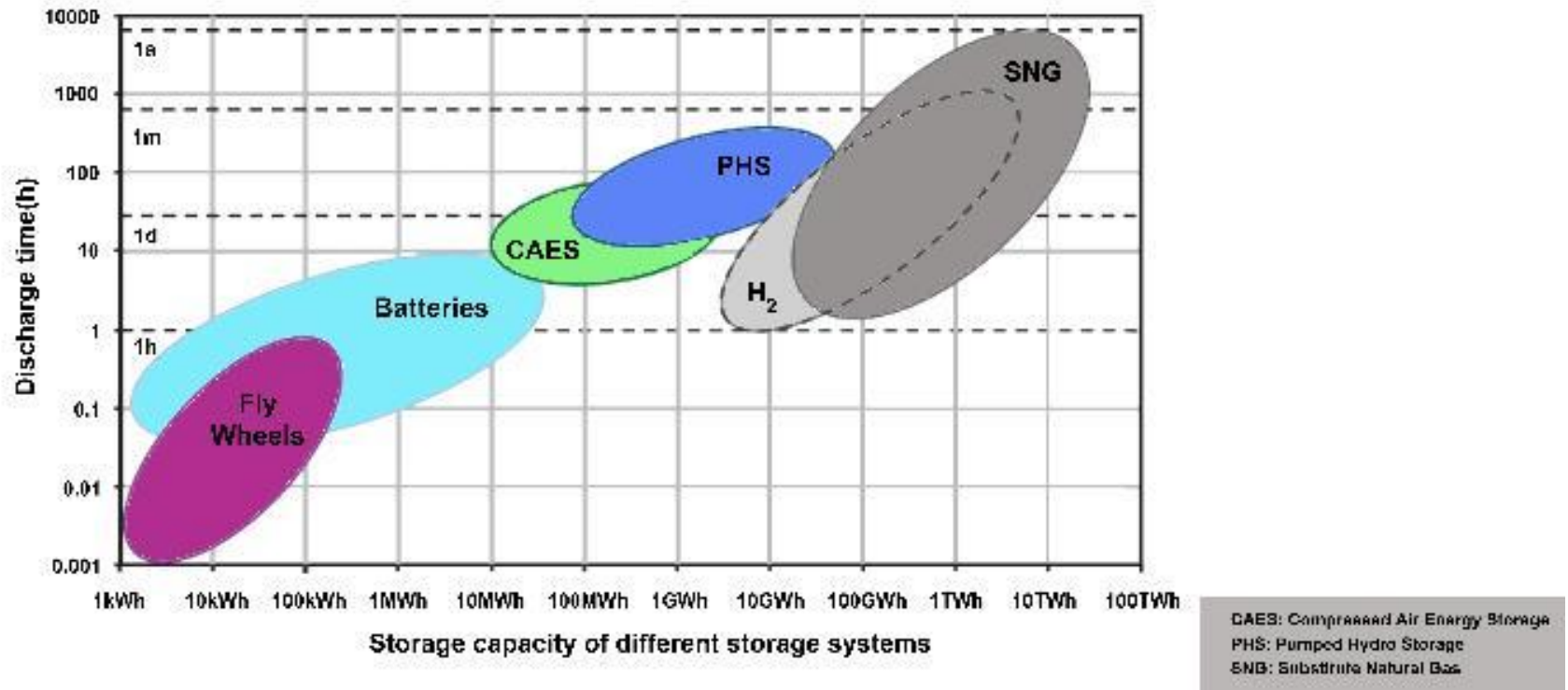
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# Passive Autocatalytic Recombiners (PARs) for H<sub>2</sub> Mitigation in Nuclear Reactor Containment

- Installed in CANDUs & PWRs
  - Licensed by Candu Energy Inc.
- 40% of French Reactors have Canadian PARs installed
- Potential for non-nuclear applications in hydrogen economy



# Hydrogen Energy Storage: A Solution To the Renewable Energy Intermittency Problem



Hydrogen gas has the largest energy content of any fuel, making it a very good 'medium' for holding and distributing energy. With the ability to hold 120MJ/kg,





# Hydrogen Storage

## Unequalled Storage Density – Utility Scale

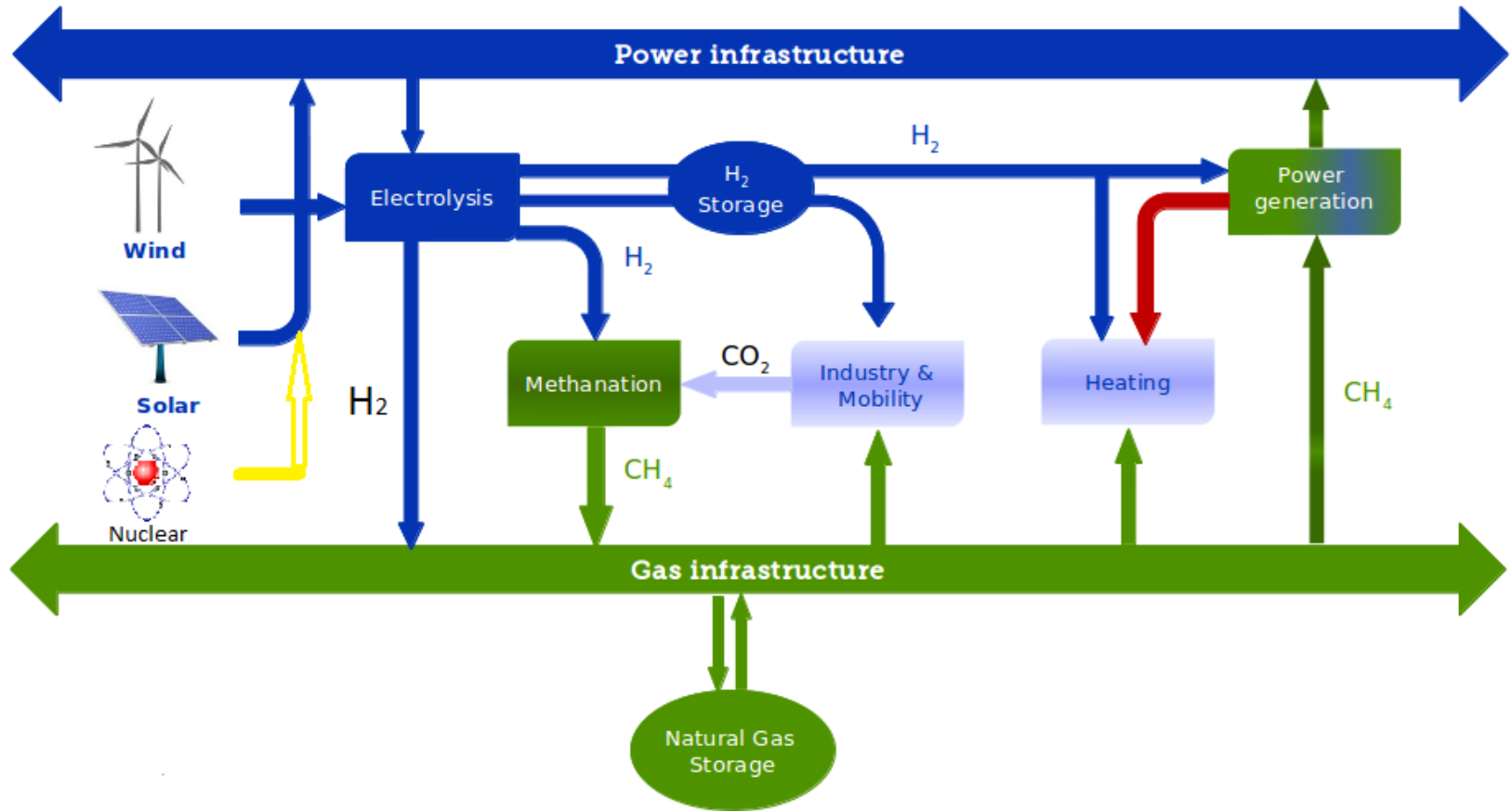
This example (actually a NG installation covering 4 acres) would contain 2.5 Gigawatt-hours of energy storage when applied to Hydrogen

- Tube trailer can deliver 4 to 6 MWh when used with fuel cell
- No leakage and no parasitic losses over time
- Minimum incremental storage capacity costs

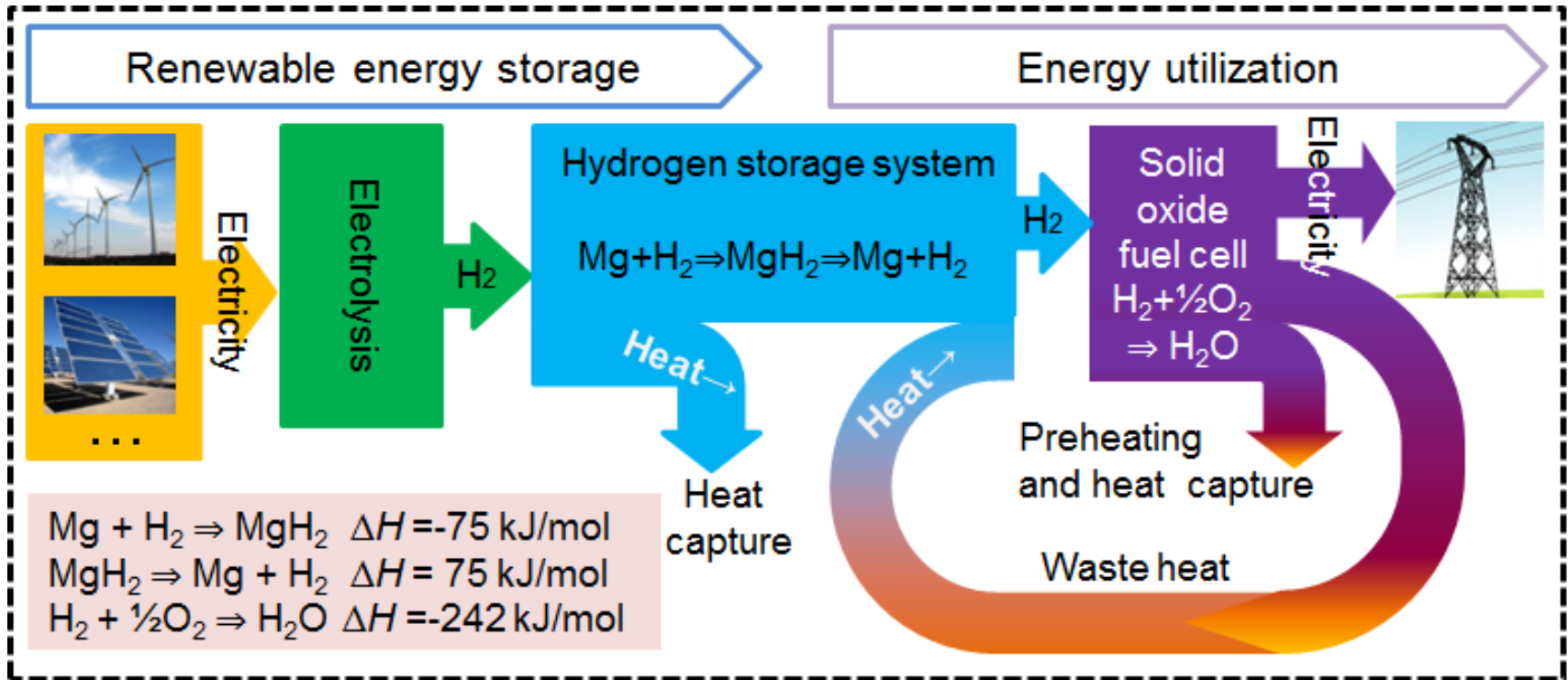




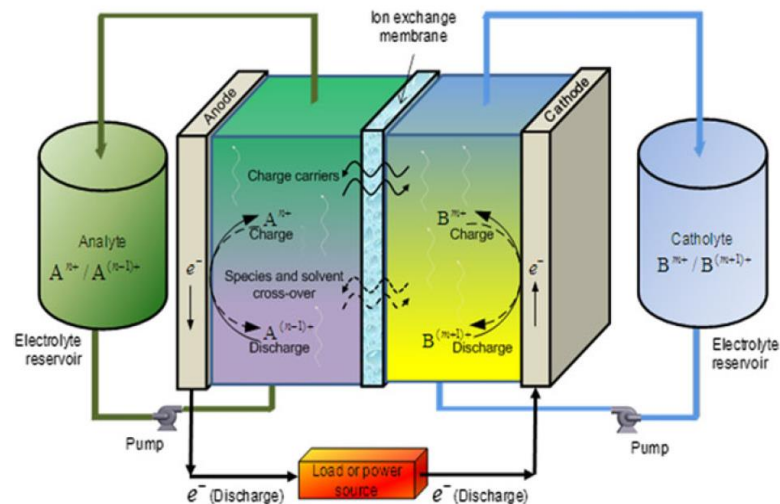
# Power-to-Gas



# Hydrogen Storage on Metals



# Flow Battery as Energy Storage Medium



- Two closed loop systems (usually liquids)
- On discharge:
  - redox species in analyte solution undergo oxidation as they are pumped through anode
  - catholyte redox species undergo reduction at cathode
  - other charge carriers diffuse through central membrane to balance charge
  - electrons flow through external circuit
- On charge: electrochemical reactions are reversed



# *Energy Scene in Canada*





# *Nuclear in Canada*

~18% electricity produced in Canada by CANDU

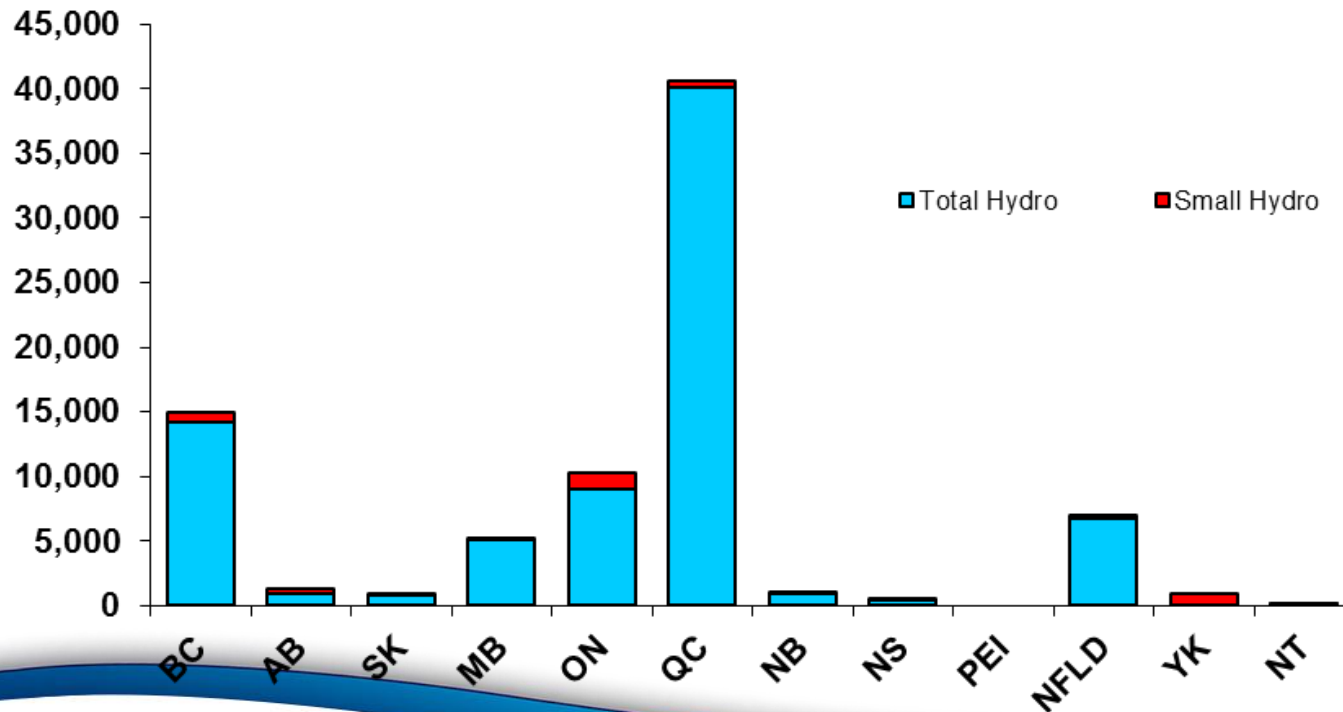
~50% electricity produced in Ontario by CANDU

Nuclear Station	Province	MWe	In service date	Operator
Pickering A	Ontario	4 x 515	1971-73	OPG
Pickering B	Ontario	4 x 516	1983-86	OPG
Darlington	Ontario	4 x 881	1990-93	OPG
Bruce A	Ontario	4 x 750	1977-79	Bruce Power
Bruce B	Ontario	4 x 860	1984-87	Bruce Power
Gentilly 2	Québec	1 x 635	1983	Hydro Québec
Point Lepreau	New Brunswick	1 x 635	1983	NB Power



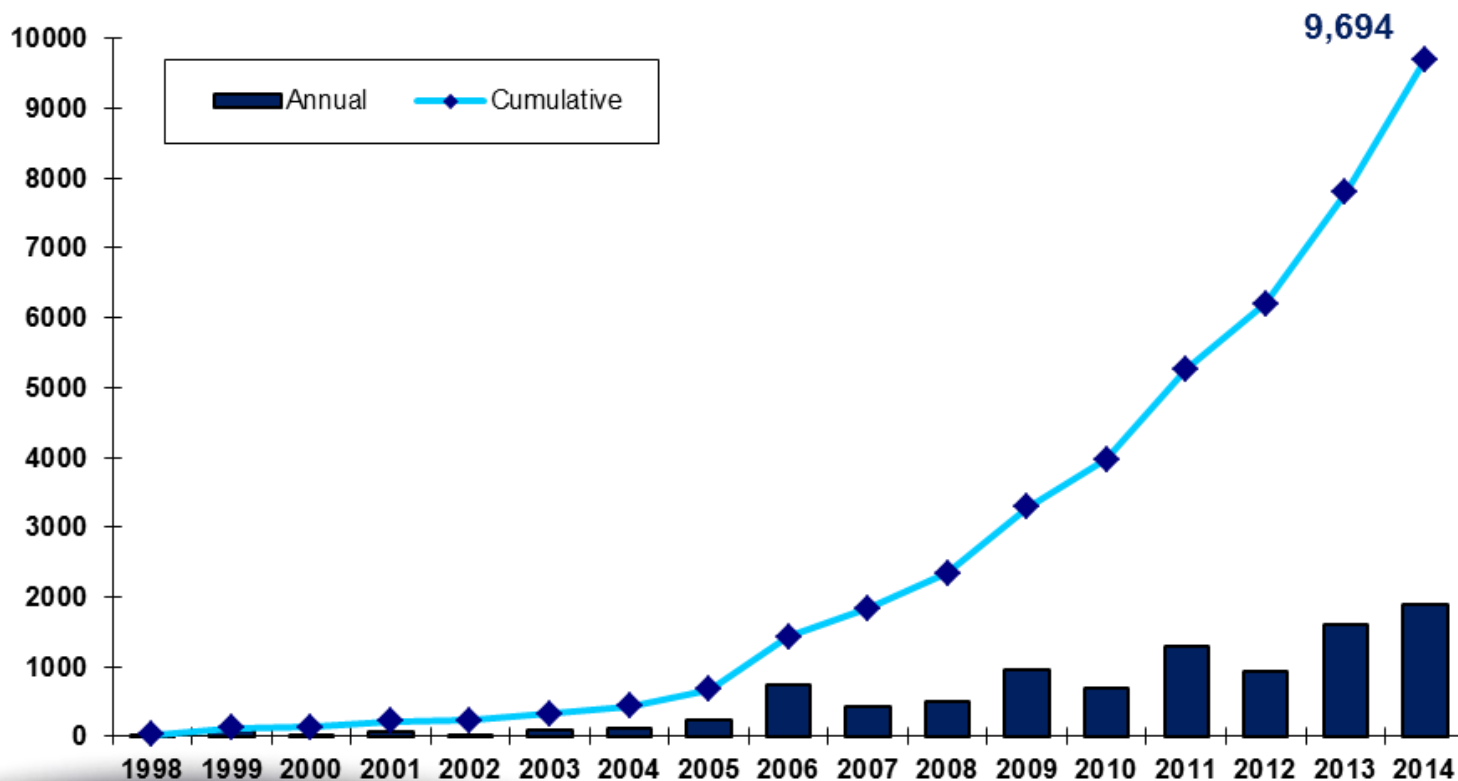
# *Installed Hydroelectric Capacity by Provinces (2014, in megawatts)*

Moving water is the most important renewable energy source in Canada, providing 59.3 per cent of Canada's electricity generation. In fact, Canada is the second largest producer of hydroelectricity in the world.



# Installed Wind Power Capacity in Canada (Megawatts)

Wind is the second most important renewable energy source in Canada. It accounts for 3.5 per cent of electricity generation in Canada.



# *Federal and Provincial Renewable Fuel Mandates (2014)*

	Renewable Alternatives to Gasoline	Renewable Alternatives to Diesel
<b>Federal</b>	5 %	2 %
<b>Provincial</b>		
British Columbia	5 %	4 %
Alberta	5 %	2 %
Saskatchewan	7.5 %	2 %
Manitoba	8.5 %	2 %
Ontario	5 %	2 %
Quebec	5 % (target only)	--





## *Energy Scene in Canada - continued*

- Canada, with its large landmass and diversified geography, has substantial renewable resources that can be used to produce energy; these resources include moving water, wind, biomass, solar, geothermal, and ocean energy.
- Canada is a world leader in the production and use of energy from renewable resources. Renewable energy sources currently provide about 18.9 per cent of Canada's total primary energy supply.
- Biomass is the third largest renewable source of Canada's electricity generation. Its share in Canada's electricity generation is 1.4 per cent.
- Wind and solar photovoltaic energy are the fastest growing sources of electricity in Canada.

**A real commitment to decarbonization – Federal and Provincial Initiatives being announced**



*Thank you*

